Information Theory and Networks Lecture 33: Information Theory, the Universe and Everything

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If you take a pack of cards as it comes from the maker and shuffle it for a few minutes, all traces of the original systematic order disappears. The order will never come back however long you shuffle. There is only one law of nature — the second law of thermodynamics — which recognises a distinction between the past and the future. Its subject is the random element in a crowd. A practical measure of the random element which can increase in the universe but never decrease is called entropy.

Arthur Eddington, The Nature of the Physical World, 1928

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Section 1

## More on the Second Law

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## The second law of thermodynamics

In a closed system, entropy cannot decrease.

- An operation is dissipative if it turns useful forms of energy into useless ones, such as heat energy
- Arrow of time implicit in this

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- most physical laws are reversible
  - ★ they don't have a "natural" direction for time
  - \* you couldn't tell if a "video" of physical events at the microscopic level was running forward or backwards
- yet most macroscopic processes are not
  - ★ largely due to the 2nd law
- 2nd law creates idea of causality?
  - ★ one things in "caused" by another in linear time
  - \* our consciousness perceives it that way because we are also subject to the second law

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Entropy is guaranteed not to decrease, but stays equal only in highest state of disorder, so it effectively increases.

Weak-nuclear force can violate time symmetry very rarely.

## Some problems • Large-scale Universe big-bang \* where does local organization come from? heat death might be OK, but big crunch reverses it ★ so long-term entropy works out the same? Black holes have no hair if black holes evaporate (Hawking's radiation) what happens to information that drops into a black hole? so black holes have entropy $S_{BH} = \frac{k_B A}{4\ell_P^2}$ where $\star$ A is the area of the event horizon ★ $\ell_P$ is the Plank length Matthew Roughan (School of Mathematical

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## The Second Law and Markov Chains

- Model an isolated system as a Markov chain
  - transitions according to physical laws governing the system
  - future of system independent of past (except through current state)
- 2nd law (in naive form) doesn't work
  - entropy can decrease

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- e.g., consider a case where the
  - \* initial distribution is uniform (max entropy)
  - $\star$  stationary distribution is non-uniform
- we could just chalk this up to Markov chains not really being covered by thermodynamics, but

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- Four different interpretations of 2nd law [CT91, pp.34-36]
  - 1 relative entropy decreases with n
  - 2 relative entropy decreases RE stationary distribution
  - entropy increases if the stationary distribution is uniform
  - 0 conditional entropy increases with n

# Information Theory More on the Second Law The Second Law and Markov Chains

2013-10-31



The problem (it seems to me) arises because a Markov chain is often a representation of a system that is closed in principle (e.g., people might not enter or leave), but not in the strict sense of thermodynamics.

## The Second Law and Markov Chains: 1 • For the Markov chain discussed above • Consider relative entropy $D(\mu_n || \mu'_n)$ where **()** $\mu_n$ and $\mu'_n$ are two probability distributions at time *n* 2 $\mu_{n+1}$ and $\mu'_{n+1}$ are corresponding distributions at time n+1Then $D(\mu_n \| \mu'_n) > D(\mu_{n+1} \| \mu'_{n+1})$ 1 relative entropy decreases with time • Think of $D(\cdot)$ as a distance 1 the two probability distributions get closer together as the system evolves 2 remember $D(\cdot)$ has a lower-bound of zero, so a limit must exist latthew Roughan (School of Mathematical : Information Theory October 31, 2013 8 / 18





### The Second Law and Markov Chains: 3

• For the Markov chain discussed above

consider the case where the stationary distribution is uniform

• We can write the relative entropy as

$$D(\mu_n \| \mu) = \log |\Omega| - H(\mu_n)$$

- **(1)** We know  $D(\mu_n || \mu)$  can't increase
- **2** so  $H(\mu_n)$  can't decrease

• So for this case  $H(\mu_n) \leq H(\mu_{n+1})$ 

- 1 this makes sense for physical systems
- **2** in equilibrium, microstates are equally likely (uniform stationary distribution)
- **③** so this kind-of handles the transition to equilibrium
- A nice example is a shuffle
  - a crude idea of a shuffle is a random permutation, with ultimately uniform distribution of all cards

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#### Theorem

A Markov chain will have uniform stationary distribution iff its probability transition matrix is doubly stochastic, i.e., all its rows and columns sum to one.

In some sense this equates to reversible physical processes, such as often considered in thermodynamics, as in this case, the transpose of the transition matrix is also a valid transition matrix.





## Landauer's Principle

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## Landauer's Principle

- Any calculation must involve some exchange of energy
  - so there is a lower bound on per bit calculation
  - any logically irreversible manipulation (e.g., erasure of a bit) is accompanied by an increase in entropy
- Landauer limit
  - minimum possible energy required to change one bit

 $= k_B T \ln 2$ 

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where  $k_B$  =Boltzmann's constant and T is temperature (in K)

- modern computers use millions of times this energy
- Practical lower bound given by T = 3K cosmic background radiation

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• Imagine a hypothetical efficient computer

- never wastes energy
- it's isolated (no energy comes in or out)
- any
  - ★ logical state (binary bits in computer) is a macrostate
  - represented by some number of microstates (physical states of electrons, magnetic particles, etc.)
- ▶ we can imagine either
  - ★ keep track of logical state
  - \star of not
- Irreversible computation
  - two or more logical states map to a single state
  - not invertible

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| $k\simeq 1.38	imes 10^{23} J/K$   |  |
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| Information Theory<br>E Landauer's Principle<br>Explanation | Explanation |
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## Explanation

(1) Don't keep track of logical state

- Irreversible calculation implies that the number of possible logical states of the computer decreases
  - in erasing a bit, we have reduced no. of states by factor of 2
  - All else equal (equal probabilities)

 $H(X) = \log_2 |\Omega|$ 

 $\star\,$  so if we reduce state space by factor of two

★ 
$$H(X)$$
 is reduced by 1 bit

- But entropy can't decrease in isolated system
  - ► there must be some other increase
  - number of physical microstates corresponding to the macrostate (or the logical bits), must have increased to compensate

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energy is dissipated into heat

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| Explanation  |
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| (2) Keep track of logical state  |
| <ul> <li>Irreversible calculation doesn't change the number of possible states<br/>(just the actual state)</li> </ul>  |
| <ul> <li>But, from previous argument the number of microstates increased</li> <li>So from the point of view of the computer's user</li> <li>entropy just increased by 1 bit</li> </ul> |
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| Information Theory | Explanation (2) Keep toxic of logical state a texture of the sumbler of possible states a texture of the sumbler of microstates increased b So from the poster of the output state wate b output posterioreand by 1 bit |
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